**Bond Enthalpy calculations (Level 2) examiners tips: Read these please!**

• a common student error is rushing these calculations and leaving out bonds eg the C - H bonds

• also using an incorrect number eg C - H instead of C = H

• C to O atoms in CO2 have a double bond O = C = O

• So, AVOID errors draw sketches showing bond types before beginning the calculation

• practise makes perfect, do all the above calculations

• then, do all similar calculations from your text books, online quizzes etc

the more calculations you can do, the easier these type of questions become

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